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catena-Poly[potassium-di- μ -aqua- μ -4-(5-tetrazolio)pyridine]

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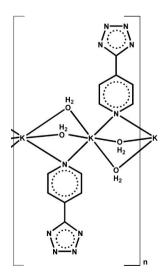
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Key indicators: single-crystal X-ray study; T = 298 K; mean σ (C–C) = 0.003 Å; R factor = 0.037; wR factor = 0.096; data-to-parameter ratio = 16.9.

The title compound, $[K(C_6H_4N_5)(H_2O)_2]_n$, was synthesized by hydrothermal reaction of KOH with 4-(5-tetrazolio)-pyridine. The K atom has a distorted octahedral coordination environment and is coordinated by two axial N atoms from the organic ligand and by four water molecules in the equatorial plane. The molecules as a whole are located on crystallographic mirror planes; the K atom is also located on an inversion center. Both the water molecules and the organic ligands act as bridges to link symmetrically the adjacent K atoms into polymeric chains parallel to the c axis. $O-H\cdots N$ hydrogen bonds involving the water O atoms and aromatic π - π interactions [centroid–centroid distance 3.80 (2) Å] between the pyridine and tetrazole rings build up an infinite three-dimensional network.

Related literature

For applications of tetrazole derivatives in coordination chemistry, see: Xiong *et al.* (2002); Wang *et al.* (2005). For the crystal structure of a related compound, see: Dai & Fu (2008).



Experimental

Crystal data

| $[K(C_6H_4N_5)(H_2O)_2]$ | $V = 991.1 (3) \text{ Å}^3$ |
|--------------------------------|---|
| $M_r = 221.27$ | Z = 4 |
| Monoclinic, C2/c | Mo $K\alpha$ radiation |
| a = 12.361 (3) Å | $\mu = 0.52 \text{ mm}^{-1}$ |
| b = 12.281 (3) Å | T = 298 K |
| c = 7.3431 (15) Å | $0.25 \times 0.15 \times 0.10 \text{ mm}$ |
| $\beta = 117.25 \ (3)^{\circ}$ | |

Data collection

| Rigaku Mercury2 diffractometer | 5056 measured reflections |
|--------------------------------------|---------------------------------------|
| Absorption correction: multi-scan | 1134 independent reflections |
| (CrystalClear; Rigaku, 2005) | 928 reflections with $I > 2\sigma(I)$ |
| $T_{\min} = 0.913, T_{\max} = 1.000$ | $R_{\rm int} = 0.027$ |
| (expected range = 0.867-0.949) | |

Refinement

| $R[F^2 > 2\sigma(F^2)] = 0.037$ | 2 restraints |
|---------------------------------|---|
| $wR(F^2) = 0.096$ | H-atom parameters constrained |
| S = 1.07 | $\Delta \rho_{\rm max} = 0.32 \text{ e Å}^{-3}$ |
| 1134 reflections | $\Delta \rho_{\min} = -0.19 \text{ e Å}^{-3}$ |
| 67 parameters | |

Table 1 Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$ | D-H | $H \cdot \cdot \cdot A$ | $D \cdot \cdot \cdot A$ | $D-\mathrm{H}\cdots A$ |
|--------------------------|------|-------------------------|-------------------------|------------------------|
| $O1W-H1WA\cdots N2^{i}$ | 0.84 | 2.01 | 2.852 (2) | 177 |
| $O1W-H1WB\cdots N3^{ii}$ | 0.88 | 1.97 | 2.831 (2) | 169 |

Symmetry codes: (i) $x - \frac{1}{2}$, $y - \frac{1}{2}$, z - 1; (ii) -x + 1, -y + 1, -z + 1.

Data collection: CrystalClear (Rigaku, 2005); cell refinement: CrystalClear; data reduction: CrystalClear; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

This work was supported by a Start-up Grant from Southeast University to Professor Ren-Gen Xiong.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2195).

References

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| supplementary m | aterials | |
|-----------------|----------|--|
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catena-Poly[potassium-di- μ -aqua- μ -4-(5-tetrazolio)pyridine]

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Comment

In the past few years, more and more people have focused on the chemistry of tetrazole derivatives because of their multiple coordination modes as ligands to metal ions and for the construction of novel metal-organic frameworks (Wang, *et al.* 2005; Xiong, *et al.* 2002). We report here the crystal structure of the title compound, tetra-aqua-bis[4-(2*H*-tetrazol-5-yl)pyridine]potassium(I).

The K atom has a distorted octahedral geometry and is coordinated by two axial pyridyl N atoms from the organic ligand and four water molecules ligands in the equatorial plane. The molecules as a whole are located on crystallographic mirror planes, the potassium ion is also located on an inversion center. Both the water molecules and the organic ligands act as bridges linking adjacent K ions into polymeric chains parallel to the c axis by covalent bonds (K—N, and K—O). The pyridine and tetrazole rings are nearly coplanar and are twisted from each other by a dihedral angle of only 12.99 (0.13) $^{\circ}$ (Fig.1). The bond distances and bond angles of the tetrazole rings are in the usual ranges (Wang, *et al.* 2005; Dai & Fu 2008).

The crystal packing (Fig. 2) is stabilized by aromatic π – π interactions between the pyridine and tetrazole rings of the neighbouring ligand systems. The centroid···centroid distance is 3.80 (2)Å (symmetry code: x, y, z+1 and x, y, z). The molecular packing is further stabilized by intermolecular O—H···N hydrogen bonds involving the aqueous O atoms. The π – π and hydrogen bonding interactions build up an infinite three-dimensional network. (Fig. 2 and Table 1).

Experimental

A mixture of 4-(2*H*-tetrazol-5-yl)pyridine (0.4 mmol) and KOH (0.4 mmol), ethanol (1 ml) and a few drops of water sealed in a glass tube was maintained at 373 K. Colorless needle crystals suitable for X-ray analysis were obtained after 3 days.

Refinement

All H atoms attached to C atoms were fixed geometrically and treated as riding with C–H = 0.93 Å (aromatic) with $U_{\rm iso}({\rm H})$ = 1.2Ueq(C). The H atoms of water molecules were located in difference Fourier maps and the O–H distances were restrained in the subsequent refinements to 0.85 Å with $U_{\rm iso}({\rm H})$ =1.5Ueq(O). In the last stage of the refinement they were treated as riding on the O atom.

supplementary materials

Figures

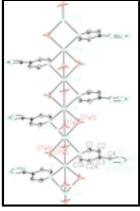


Fig. 1. A view of the title compound with the atomic numbering scheme. Displacement ellipsoids were drawn at the 30% probability level.



Fig. 2. The crystal packing of the title compound viewed along the *c* axis showing the three dimensionnal network (dashed lines). Hydrogen atoms not involved in hydrogen bonding have been omitted for clarity.

catena-Poly[potassium(I)-di-µ-aqua-µ-4-(5-tetrazolio)pyridine]

Crystal data

 $[K(C_6H_4N_5)(H_2O)_2]$ $F_{000} = 456$ $M_r = 221.27$ $D_{\rm x} = 1.483~{\rm Mg~m}^{-3}$ Mo $K\alpha$ radiation Monoclinic, C2/c $\lambda = 0.71073 \text{ Å}$ Hall symbol: -C 2yc Cell parameters from 1134 reflections a = 12.361 (3) Å $\theta = 3.3-27.5^{\circ}$ b = 12.281 (3) Å $\mu = 0.52 \text{ mm}^{-1}$ T = 298 Kc = 7.3431 (15) Å $\beta = 117.25 (3)^{\circ}$ Needle, colorless $V = 991.1 (3) \text{ Å}^3$ $0.25\times0.15\times0.10~mm$ Z = 4

Data collection

Rigaku Mercury2 (2× 2 bin mode) 1134 independent reflections diffractometer Radiation source: fine-focus sealed tube 928 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.027$ Monochromator: graphite $\theta_{\text{max}} = 27.5^{\circ}$ Detector resolution: 13.6612 pixels mm⁻¹ $\theta_{min} = 3.3^{\circ}$ T = 298 K $h = -16 \rightarrow 16$ ω scans Absorption correction: multi-scan $k = -15 \rightarrow 15$ (CrystalClear; Rigaku, 2005) $T_{\min} = 0.913, T_{\max} = 1.000$ $l = -9 \rightarrow 9$

5056 measured reflections

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.037$

 $wR(F^2) = 0.096$

 $wR(F^2) = 0.096$

S = 1.07

1134 reflections

67 parameters

2 restraints
Primary atom site location: structure-invariant direct

methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring

sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_0^2) + (0.039P)^2 + 0.7641P]$

where $P = (F_0^2 + 2F_c^2)/3$

 $(\Delta/\sigma)_{\text{max}} < 0.001$

 $\Delta \rho_{\text{max}} = 0.32 \text{ e Å}^{-3}$

 $\Delta \rho_{min} = -0.19 \text{ e Å}^{-3}$

Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\mathring{A}^2)

| | x | y | Z | $U_{\rm iso}*/U_{\rm eq}$ |
|------|--------------|--------------|------------|---------------------------|
| K1 | 0.5000 | 0.0000 | 0.5000 | 0.0467(2) |
| C4 | 0.5000 | 0.5325 (2) | 0.7500 | 0.0365 (6) |
| C3 | 0.5000 | 0.4124 (2) | 0.7500 | 0.0358 (5) |
| N2 | 0.60000 (14) | 0.59305 (13) | 0.8162 (3) | 0.0468 (4) |
| C2 | 0.60283 (18) | 0.35391 (16) | 0.7821 (3) | 0.0459 (5) |
| H2 | 0.6742 | 0.3897 | 0.8050 | 0.055* |
| N3 | 0.55965 (15) | 0.69599 (13) | 0.7896 (3) | 0.0530 (5) |
| N1 | 0.5000 | 0.18446 (19) | 0.7500 | 0.0522 (6) |
| C1 | 0.5980(2) | 0.24208 (17) | 0.7797 (4) | 0.0534 (5) |
| H1 | 0.6678 | 0.2041 | 0.8000 | 0.064* |
| O1W | 0.34559 (12) | 0.08972 (11) | 0.1308 (2) | 0.0516 (4) |
| H1WA | 0.2731 | 0.0885 | 0.0381 | 0.077* |
| H1WB | 0.3654 | 0.1588 | 0.1421 | 0.077* |

supplementary materials

| Atomic displacement parameters (\mathring{A}^2) | | | | | | | |
|---|------------------|-------------|----------------------|----------------------------|-------------|-------------|--|
| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} | |
| K1 | 0.0545 (4) | 0.0460 (4) | 0.0403 (3) | -0.0005 (3) | 0.0221 (3) | -0.0025 (3) | |
| C4 | 0.0345 (13) | 0.0344 (12) | 0.0323 (13) | 0.000 | 0.0080 (11) | 0.000 | |
| C3 | 0.0381 (13) | 0.0334 (13) | 0.0296 (12) | 0.000 | 0.0102 (10) | 0.000 | |
| N2 | 0.0383 (9) | 0.0337 (8) | 0.0522 (10) | -0.0025 (7) | 0.0067 (7) | -0.0016 (7) | |
| C2 | 0.0392 (10) | 0.0406 (10) | 0.0536 (12) | -0.0008(8) | 0.0176 (9) | -0.0029 (9) | |
| N3 | 0.0503 (9) | 0.0340 (8) | 0.0541 (11) | -0.0042 (7) | 0.0062 (8) | -0.0020 (7) | |
| N1 | 0.0602 (16) | 0.0343 (12) | 0.0524 (15) | 0.000 | 0.0174 (13) | 0.000 | |
| C1 | 0.0514 (12) | 0.0416 (11) | 0.0596 (13) | 0.0090 (9) | 0.0189 (10) | -0.0026 (9) | |
| O1W | 0.0341 (7) | 0.0376 (7) | 0.0667 (10) | 0.0014 (6) | 0.0088 (7) | 0.0026 (6) | |
| Geometric para | meters (Å, °) | | | | | | |
| K1—O1W ⁱ | | 2.7309 (16) | C3- | —C2 ^{iv} | 1.38 | 33 (2) | |
| K1—O1W ⁱⁱ | | 2.7309 (16) | C3- | —C2 | 1.38 | 33 (2) | |
| K1—O1W | | 2.7330 (17) | N2- | —N3 | 1.34 | 10 (2) | |
| K1—O1W ⁱⁱⁱ | | 2.7330 (17) | C2- | —C1 | 1.375 (3) | | |
| K1—N1 ⁱⁱⁱ | | 2.9159 (18) | C2- | —H2 | 0.9300 | | |
| K1—N1 | | 2.9159 (18) | N3- | -N3 ^{iv} | 1.314 (3) | | |
| K1—C1 ⁱⁱⁱ | | 3.499 (2) | N1—C1 | | 1.332 (3) | | |
| K1—C1 | | 3.499 (2) | N1—C1 ^{iv} | | 1.332 (3) | | |
| K1—K1 ⁱⁱ | | 3.6716 (7) | N1—K1 ^{iv} | | 2.91 | 59 (18) | |
| K1—K1 ^{iv} | | 3.6716 (8) | C1—H1 | | 0.93 | 800 | |
| K1—H1WB | | 3.0780 | O1W—K1 ⁱⁱ | | 2.73 | 809 (16) | |
| C4—N2 | | 1.329 (2) | O1W—H1WA | | 0.84 | 112 | |
| C4—N2 ^{iv} | | 1.329 (2) | O1W—H1WB | | 0.8765 | | |
| C4—C3 | | 1.475 (3) | | | | | |
| O1W ⁱ —K1—O1 | W^{ii} | 180.00(3) | C1 ⁱ | ii—K1—K1 ^{iv} | 115 | 36 (4) | |
| O1W ⁱ —K1—O1 | W | 103.20 (5) | C1- | —К1—К1 ^{iv} | 64.64 (4) | | |
| O1W ⁱⁱ —K1—O1 | W | 76.80 (5) | K1 ⁱ | ii—K1—K1 ^{iv} | 180.0 | | |
| O1W ⁱ —K1—O1 | W ⁱⁱⁱ | 76.80 (5) | O1' | O1W ⁱ —K1—H1WB | | 4 | |
| O1W ⁱⁱ —K1—O1 | W^{iii} | 103.20 (5) | 01 | O1W ⁱⁱ —K1—H1WB | | 68.6 | |
| O1W—K1—O1V | $W^{ m iii}$ | 180.0 | O1' | W—K1—H1WB | 16.0 |) | |
| O1W ⁱ —K1—N1 ⁱ | iii | 96.28 (4) | O1' | W ⁱⁱⁱ —K1—H1WB | 164 | .0 | |
| O1W ⁱⁱ —K1—N1 | iii | 83.72 (4) | N1 ⁱ | iii—K1—H1WB | 96.4 | ŀ | |
| O1W—K1—N1 ⁱⁱ | ii | 83.68 (4) | N1- | —K1—Н1WВ | 83.6 | 5 | |
| O1W ⁱⁱⁱ —K1—N1 | l ⁱⁱⁱ | 96.32 (4) | C1 ⁱ | ⁱⁱ —K1—H1WB | 97.5 | ; | |
| O1W ⁱ —K1—N1 | | 83.72 (4) | C1- | —K1—H1WB | 82.5 | ; | |
| O1W ⁱⁱ —K1—N1 | | 96.28 (4) | K1 ⁱ | i—K1—H1WB | 52.7 | 1 | |
| O1W—K1—N1 | | 96.32 (4) | K1 ⁱ | iv—K1—H1WB | 127 | .3 | |
| O1W ⁱⁱⁱ —K1—N | 1 | 83.68 (4) | N2- | —C4—N2 ^{iv} | 112 | .0 (2) | |

supplementary materials

| N1 ⁱⁱⁱ —K1—N1 | 180.0 | N2—C4—C3 | 124.01 (11) | | | |
|---|---|------------------------------|-------------|--|--|--|
| O1W ⁱ —K1—C1 ⁱⁱⁱ | 75.74 (5) | N2 ^{iv} —C4—C3 | 124.01 (11) | | | |
| O1W ⁱⁱ —K1—C1 ⁱⁱⁱ | 104.26 (5) | C2 ^{iv} —C3—C2 | 117.4 (2) | | | |
| O1W—K1—C1 ⁱⁱⁱ | 82.15 (5) | C2 ^{iv} —C3—C4 | 121.30 (12) | | | |
| O1W ⁱⁱⁱ —K1—C1 ⁱⁱⁱ | 97.85 (5) | C2—C3—C4 | 121.30 (12) | | | |
| N1 ⁱⁱⁱ —K1—C1 ⁱⁱⁱ | 21.60 (4) | C4—N2—N3 | 104.64 (16) | | | |
| N1—K1—C1 ⁱⁱⁱ | 158.40 (4) | C1—C2—C3 | 119.1 (2) | | | |
| O1W ⁱ —K1—C1 | 104.26 (5) | C1—C2—H2 | 120.5 | | | |
| O1W ⁱⁱ —K1—C1 | 75.74 (5) | C3—C2—H2 | 120.5 | | | |
| O1W—K1—C1 | 97.85 (5) | N3 ^{iv} —N3—N2 | 109.38 (10) | | | |
| O1W ⁱⁱⁱ —K1—C1 | 82.15 (5) | C1—N1—C1 ^{iv} | 115.8 (2) | | | |
| N1 ⁱⁱⁱ —K1—C1 | 158.40 (4) | C1—N1—K1 | 104.69 (11) | | | |
| N1—K1—C1 | 21.60 (4) | C1 ^{iv} —N1—K1 | 124.89 (11) | | | |
| C1 ⁱⁱⁱ —K1—C1 | 180.00 (9) | C1—N1—K1 ^{iv} | 124.89 (11) | | | |
| O1W ⁱ —K1—K1 ⁱⁱ | 132.19 (4) | $C1^{iv}$ — $N1$ — $K1^{iv}$ | 104.69 (11) | | | |
| O1W ⁱⁱ —K1—K1 ⁱⁱ | 47.81 (4) | K1—N1—K1 ^{iv} | 78.04 (6) | | | |
| O1W—K1—K1 ⁱⁱ | 47.76 (3) | N1—C1—C2 | 124.3 (2) | | | |
| O1W ⁱⁱⁱ —K1—K1 ⁱⁱ | 132.24 (3) | N1—C1—K1 | 53.71 (10) | | | |
| N1 ⁱⁱⁱ —K1—K1 ⁱⁱ | 50.98 (3) | C2—C1—K1 | 149.08 (16) | | | |
| N1—K1—K1 ⁱⁱ | 129.02 (3) | N1—C1—H1 | 117.8 | | | |
| C1 ⁱⁱⁱ —K1—K1 ⁱⁱ | 64.64 (4) | C2—C1—H1 | 117.8 | | | |
| C1—K1—K1 ⁱⁱ | 115.36 (4) | K1—C1—H1 | 73.4 | | | |
| O1W ⁱ —K1—K1 ^{iv} | 47.81 (4) | K1 ⁱⁱ —O1W—K1 | 84.44 (4) | | | |
| O1W ⁱⁱ —K1—K1 ^{iv} | 132.19 (4) | K1 ⁱⁱ —O1W—H1WA | 111.4 | | | |
| O1W—K1—K1 ^{iv} | 132.24 (3) | K1—O1W—H1WA | 142.8 | | | |
| O1W ⁱⁱⁱ —K1—K1 ^{iv} | 47.76 (3) | K1 ⁱⁱ —O1W—H1WB | 102.4 | | | |
| $N1^{iii}$ — $K1$ — $K1^{iv}$ | 129.02 (3) | K1—O1W—H1WB | 105.0 | | | |
| N1—K1—K1 ^{iv} | 50.98 (3) | H1WA—O1W—H1WB | 104.0 | | | |
| Symmetry codes: (i) x , $-y$, $z+1/2$; (ii) $-$ | Symmetry codes: (i) x , $-y$, $z+1/2$; (ii) $-x+1$, y , $-z+1/2$; (iii) $-x+1$, $-y$, $-z+1$; (iv) $-x+1$, y , $-z+3/2$. | | | | | |
| | | | | | | |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H··· <i>A</i> | <i>D</i> —H | $H\cdots A$ | D··· A | <i>D</i> —H <i>···A</i> |
|-----------------------------|-------------|-------------|-----------|-------------------------|
| O1W—H1WA···N2 ^v | 0.84 | 2.01 | 2.852 (2) | 177 |
| O1W—H1WB···N3 ^{vi} | 0.88 | 1.97 | 2.831 (2) | 169 |

Symmetry codes: (v) x-1/2, y-1/2, z-1; (vi) -x+1, -y+1, -z+1.

Fig. 1

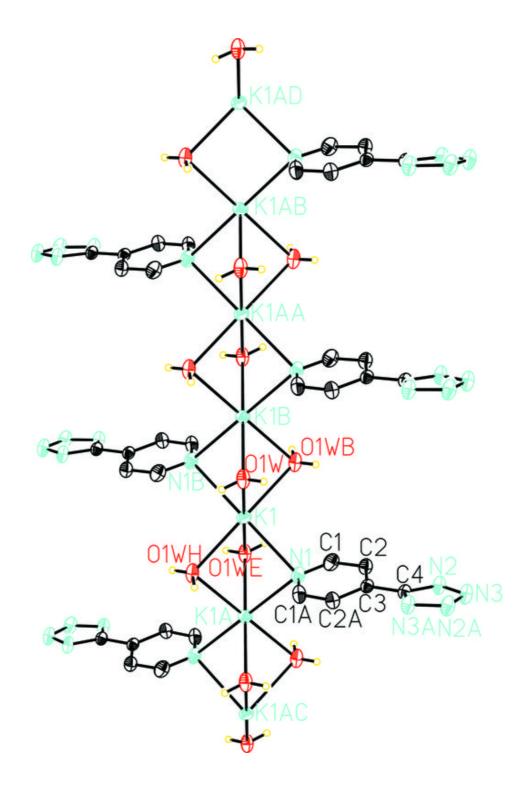


Fig. 2

